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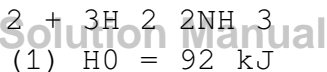
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1 Chemical
reactions 1.1
Rate of reaction
and dependence
on temperature
We will once
again look at
the formation of
ammonia (NH_3)
from nitrogen
and hydrogen
(see section
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equilibrium of

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$\text{mol S}_0 = 192 \text{ J}$
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recall that $G_0 =$
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kJ mol⁻¹ + (298 K)
0.192 kJ mol⁻¹ K =
35 kJ mol⁻¹

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number 18. It is in group 18 of the periodic

table and is a noble gas. Argon is the third most common gas in the Earth's atmosphere, at 0.934% (9,340

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An apparatus for growing organisms

(yeast,

bacteria, or animal cells)

under controlled conditions. Used in industrial processes to produce

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used to convert
raw materials
into useful
byproducts such
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