

Chemistry Chapter 3 Scientific Measurement

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Chemistry Chapter 3-Scientific Measurement. STUDY. PLAY. measurement. a quantitative description that includes both a number and a unit. Scientific notation. an expression of numbers in the form $m \times 10^n$ where m is equal to or greater than 1 and less than 10 and n is an integer. accuracy.

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PEP -- Chemistry 1 Chemistry/ PEP Name: _____ Date: _____ Chapter 3 -- Scientific Measurement Chapter 3: 1 -- 24, 26 - 28, 32, 34, 38, 40, 42, 46, 51, 56, 57, 62, 85, 87 (39 total) Section Review 3.1 1. a. What is the difference between a qualitative measurement and a quantitative measurement? b. How is a number converted to scientific ...

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• In chemistry, the meanings of accuracy and precision are quite different. • Accuracy is a measure of how close a measurement comes to the actual or true value of whatever is measured. • Precision is a measure of how close a series of measurements are to one another, irrespective of the actual value. Accuracy, Precision, and Error

3.1 Using and Expressing Measurements >
Section 3: Scientific Notation. The study of chemistry can involve numbers that are very large. It can also involve numbers that are very small. Writing out such numbers and using them in their long form is problematic, because we would spend far too much time writing zeroes, and we would probably make a lot of mistakes!

Chapter 1: Measurements in Chemistry -- Chemistry
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Title: Chapter 3 Scientific Measurement 1 Chapter 3Scientific Measurement. Charles Page High School ; Dr. Stephen L. Cotton; 2 Section 3.1The Importance of Measurement. OBJECTIVES ; Distinguish between quantitative and qualitative measurements. 3 Section 3.1The Importance of Measurement. OBJECTIVES ; Convert measurements to scientific notation. 4 Measurements

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