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CHEMISTRY 101: Scientific Measurements

Chapter 3 – Scientific Measurement Chapter 3: 1 – 24, 26 - 28, 32, 34, 38, 40, 42, 46, 51, 56, 57, 62, 85, 87 (39 total) Section Review 3.1 1. a. What is the difference between a qualitative measurement? Qualitative measurement? Qualitative measurements are expressed in descriptive, non-numerical form, whereas

Round off each answer correctly. A) 8.7g + 15.43g + 19g = 43.13g. B) 5.40 m x 3.21 m x 1.871 m = 32.431 914 m³. A) 43g. B) 32.2 m³. the measure of how close a measurement comes to the actual or true value of whatever is measured. Accuracy. a measure of how close a series of measurements are to one another. Precision.

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Scientific Measurement Chapters 3 Assignment & Problem Set Chapter 3 – Scientific Measurement. Jennie L. Borders. Section 3.1 – Measurements and Their Uncertainty. A measurement is a quantity that has both a number and a unit. The unit typically used in the sciences are those of the International System of Measurements (SI). In scientific notation, a given number is written as the product of two numbers: a coefficient and 10 raised to a power.

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Chapter 3 - Scientific Measurement - 3.1 Using and Expressing Measurements - Sample Problem 3.4 - Page 69: 6 Answer A) 8.71 x 10 meters B) 4.36 x 10^8 meters C) 1.55 x 10^-2 meters D) 9.01 x 10^3 meters E) 1.78 x 10^-3 meters F) 6.30 x 10^2 meters

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