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~~CHAPTER 2 Acids, Bases and Salts~~

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~~Acid + base → salt + water + heat. H<sub>2</sub>SO<sub>4</sub> + MgO → MgSO<sub>4</sub> + H<sub>2</sub>O 2HCl + Mg(OH)<sub>2</sub> → MgCl<sub>2</sub> + 2H<sub>2</sub>O. 2. Reaction of non-metal oxides with bases. Non-metal oxides are acidic in nature Base + Non-metal oxide → salt + water + heat. 2NaOH + CO<sub>2</sub> → Na<sub>2</sub>CO<sub>3</sub> + H<sub>2</sub>O. To know more about Properties of Acids and Bases, visit here. Water Acids and bases in water~~

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~~salts (handouts) Chapter 19...Acids, Bases, and Salts ACID-BASE THEORIES Acids and bases are all around us and part of our everyday life (ex. bodily functions, vinegar, carbonated drinks, citrus fruits, car batteries, antacids, cleaning products, etc.). Chapter 19 Acids Bases And Salts Answer Key Salts . When acid and base neutralize, salts are formed.~~

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~~acids bases and salts class 7 Science Chapter 5: In today's Class 7 Science Sprint session in Chapter 5, we discuss acids bases and salts class 7. The aspirants...~~

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~~Chapter 19 "Acids, Bases, and Salts". Use these activities to study the vocabulary and major concepts presented in this chapter. compounds derived from the reaction of a strong base with a weak acid or a strong acid with a weak base or a solution of a weak base and one of its salts.~~

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