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the metal to be protected to a more easily corroded "sacrificial metal" to act as the anode. The sacrificial metal then corrodes instead of the protected metal. Cathodic protection - Wikipedia Cathodic protection ; Guide ...

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The Corrosion

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Handbook combines

the author's previous

three works,

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in one place, the most

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available to the
engineer or student.
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added a tremendous
and exhaustive list of
questions and
answers based on the
text, which can be
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Cathodic protection (CP) is a technique used to control the corrosion of a metal surface by making it the cathode of an electrochemical cell. A simple method of protection connects the metal to be protected to a more easily corroded "sacrificial metal" to act as the anode. The

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sacrificial metal then
corrodes instead of
the protected metal.

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Cathodic protection - Wikipedia

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Guide for Oil and Gas
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the necessary reports,
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and critical standards

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needed on various oil and gas equipment, structures, tanks, and pipelines. Demands in the cathodic protection market have driven development for better devices and methods, helping to prolong the equipment and pipeline ...

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The cathodic protection of plain carbon and low-alloy steels can be achieved with galvanic anodes of zinc, aluminum, or magnesium. For materials with relatively more

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positive protection potentials, galvanic anodes of iron or of activated lead can be used.

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This Guide describes the basic principles of cathodic protection, the areas of use, and

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the general factors to be considered in the choice and design of a system. It gives a basic introduction and simple technical data on cathodic protection.

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The Corrosion
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Handbook combines the author's previous three works, Corrosion Chemistry, Cathodic Protection, and Corrosion Engineering to offer, in one place, the most comprehensive and thorough work available to the engineer or student. The author has also added a tremendous

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and exhaustive list of questions and answers based on the text, which can be used ...

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Cathodic protection (CP) is one of the few methods that

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successfully mitigates corrosion. It can be applied in any situation where the environment surrounding the metal acts as a conductor for electric current.

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minimizing the difference in potential between anode and cathode. This is achieved by applying a current to the structure to be protected (such as a pipeline) from some outside source.

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